

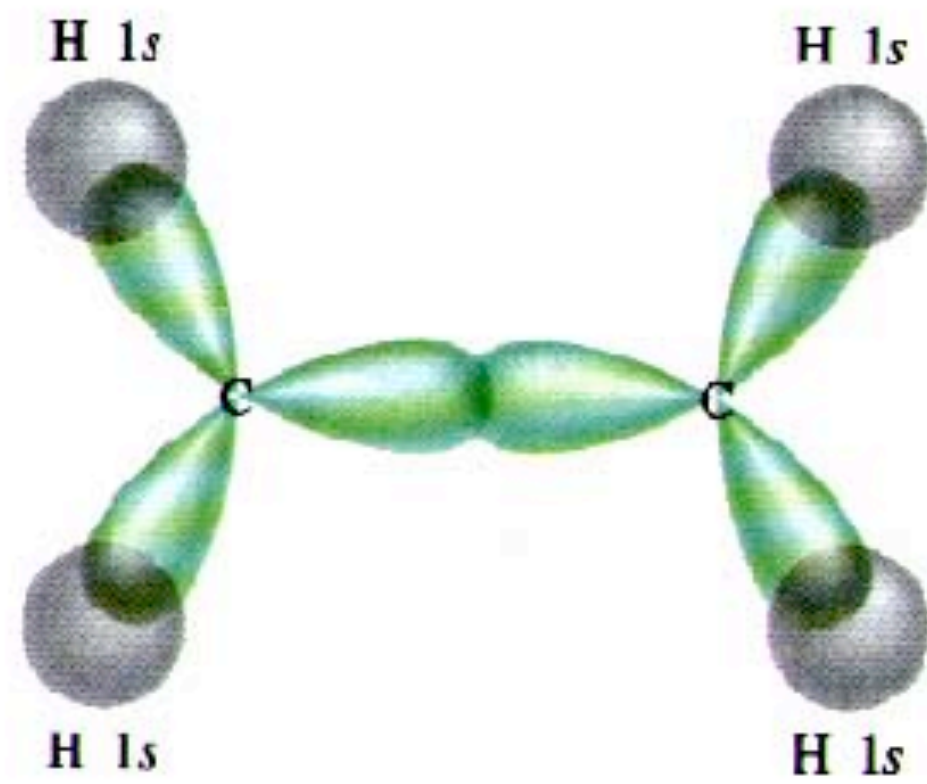
# Analysis tools for electronic structure and bonding properties

Alberto García

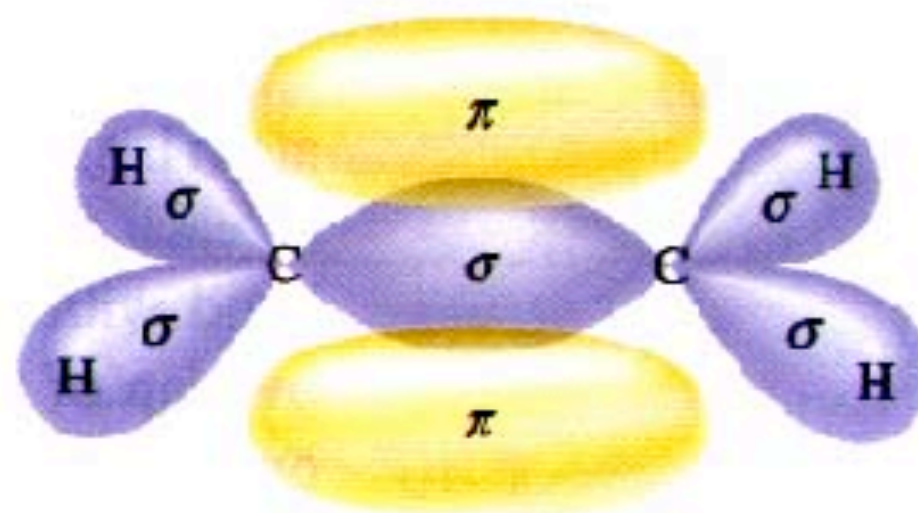
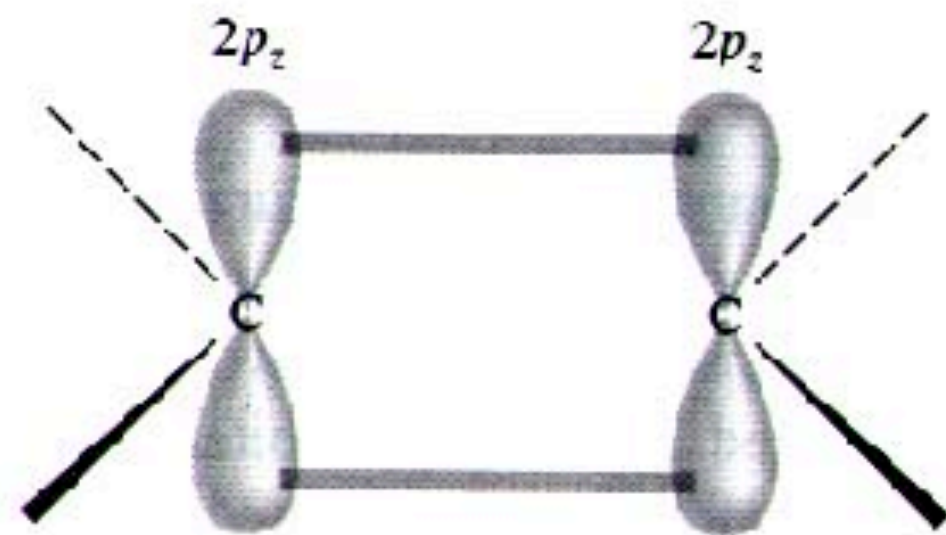
Institut de Ciència de Materials de Barcelona (CSIC)

# Why atomic orbitals?

- “Atoms” are a very good first approximation.
- The size of the basis is relatively small.
- Most of the language of the chemical bond is based on atomic orbitals.



(a)



$$\psi_i(\mathbf{r}) = \sum_{\mu} \phi_{\mu}(\mathbf{r}) c_{\mu i},$$

$$\sum_{\nu\beta} (H_{\mu\nu}^{\alpha\beta} - E_i S_{\mu\nu} \delta^{\alpha\beta}) c_{\nu i}^{\beta} = 0$$

Generalized  
eigenvalue problem

$$H_{\mu\nu}^{\alpha\beta} = \langle \phi_{\mu} | \hat{T} + \hat{V}^{KB} + V^{NA}(\mathbf{r}) + \delta V^H(\mathbf{r}) + V_{XC}^{\alpha\beta}(\mathbf{r}) | \phi_{\nu} \rangle$$

$$S_{\mu\nu} = \langle \phi_{\mu} | \phi_{\nu} \rangle$$

$$\rho(\mathbf{r}) = \sum_{\mu\nu} \rho_{\mu\nu} \phi_{\nu}^*(\mathbf{r}) \phi_{\mu}(\mathbf{r})$$

$$E^{BS} = \sum_i n_i \langle \psi_i | \hat{H} | \psi_i \rangle = \sum_{\mu\nu} H_{\mu\nu} \rho_{\nu\mu} = \text{Tr}(H\rho)$$

$$\rho_{\mu\nu} = \sum_i c_{\mu i} n_i c_{i\nu}$$

Density matrix

$$E^{BS} = \sum_i n_i \langle \psi_i | \hat{H} | \psi_i \rangle = \sum_{\mu\nu} H_{\mu\nu} \rho_{\nu\mu} = \text{Tr}(H\rho)$$

$$N = \sum_{\mu\nu} S_{\mu\nu} \rho_{\nu\mu} = \text{Tr}(S\rho)$$

$$\rho_{\mu\nu} = \sum_i c_{\mu i} n_i c_{i\nu}$$

$$\rho_{\mu\nu} = \sum_i c_{\mu i} n_i c_{i\nu} = \int_0^{\epsilon_F} d\epsilon \sum_i c_{\mu i} c_{i\nu} \delta(\epsilon - \epsilon_i)$$

$$N = \int_0^{\epsilon_F} d\epsilon \sum_i \sum_{\mu} \sum_{\nu} c_{\mu i} c_{i\nu} S_{\mu\nu} \delta(\epsilon - \epsilon_i)$$

$$N = \int_0^{\epsilon_F} d\epsilon \sum_i \sum_{\mu} \sum_{\nu} c_{\mu i} c_{i\nu} S_{\mu\nu} \delta(\epsilon - \epsilon_i)$$

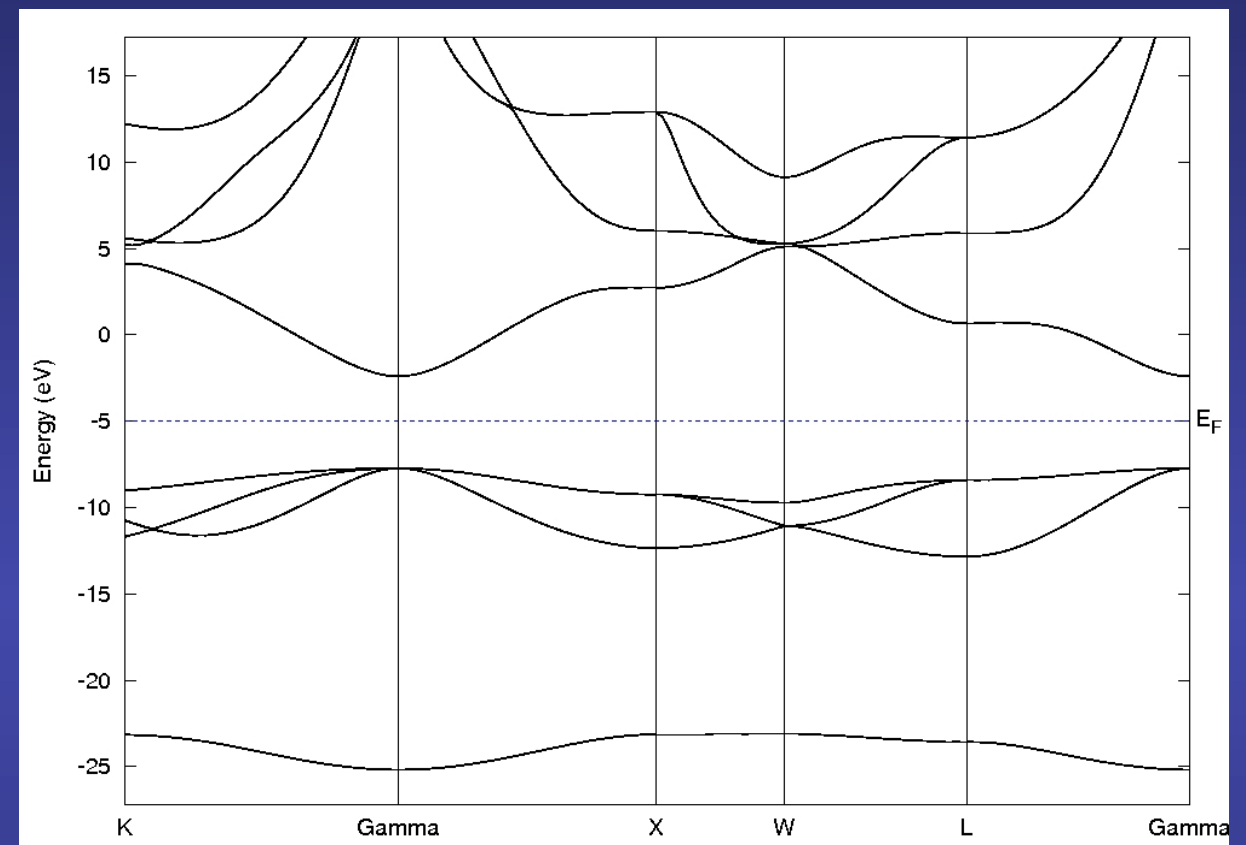
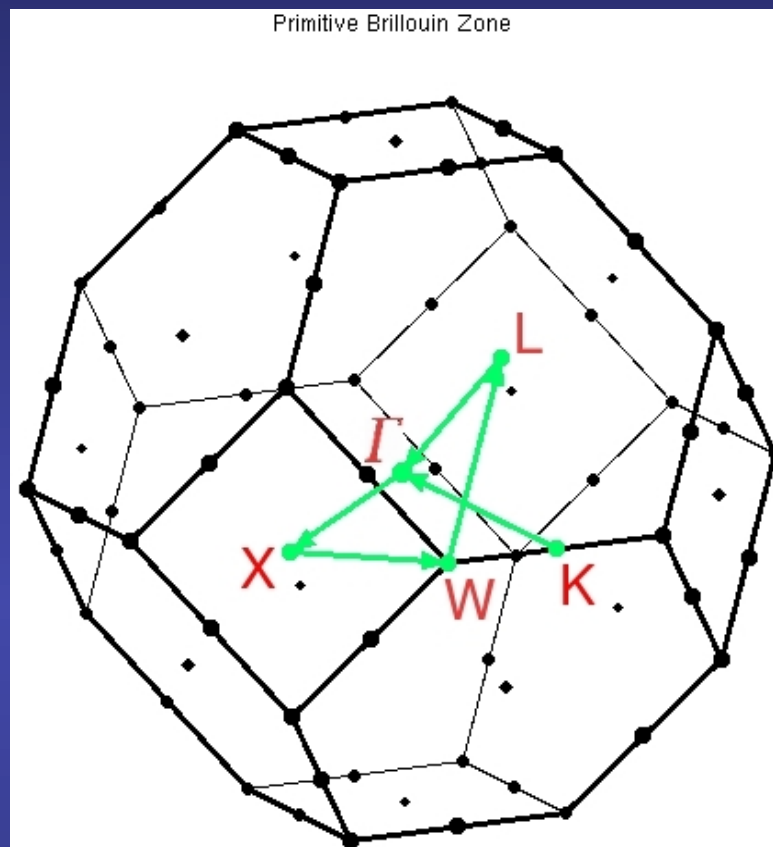
Density of states:

$$DOS(\epsilon) = g(\epsilon) = \sum_i \sum_{\mu} \sum_{\nu} c_{\mu i} c_{i\nu} S_{\mu\nu} \delta(\epsilon - \epsilon_i)$$

Density of states projected on orbital  $\mu$ :

$$g_{\mu}(\epsilon) = \sum_i \sum_{\nu} c_{\mu i} c_{i\nu} S_{\mu\nu} \delta(\epsilon - \epsilon_i)$$

# Band structure of an ionic solid: The case of MgO



```
%block kgrid_Monkhorst_Pack
  6  0  0  0.5
  0  6  0  0.5
  0  0  6  0.5
%endblock kgrid_Monkhorst_Pack
...
MeshCutoff                200 Ry
...
COOP.write T               # NOTE
```

## New H,S, and wavefunction files produced

```
-rw-r-----  1 ag  ag  397564 Jun 10 17:59 MgO.HSX
-rw-r-----  1 ag  ag  358248 Jun 10 17:59 MgO.fullBZ.WFSX
```



# Input file for (P)DOS processing by mprop

MgO	SystemLabel
DOS	Keyword
PDOS_Mg	Curve Label
Mg	Orbital spec
PDOS_O	...
O	
PDOS_O_2s	Curve Label
O_2s	Orbital spec
PDOS_O_2p	...
O_2p	
PDOS_Mg_3s	
Mg_3s	

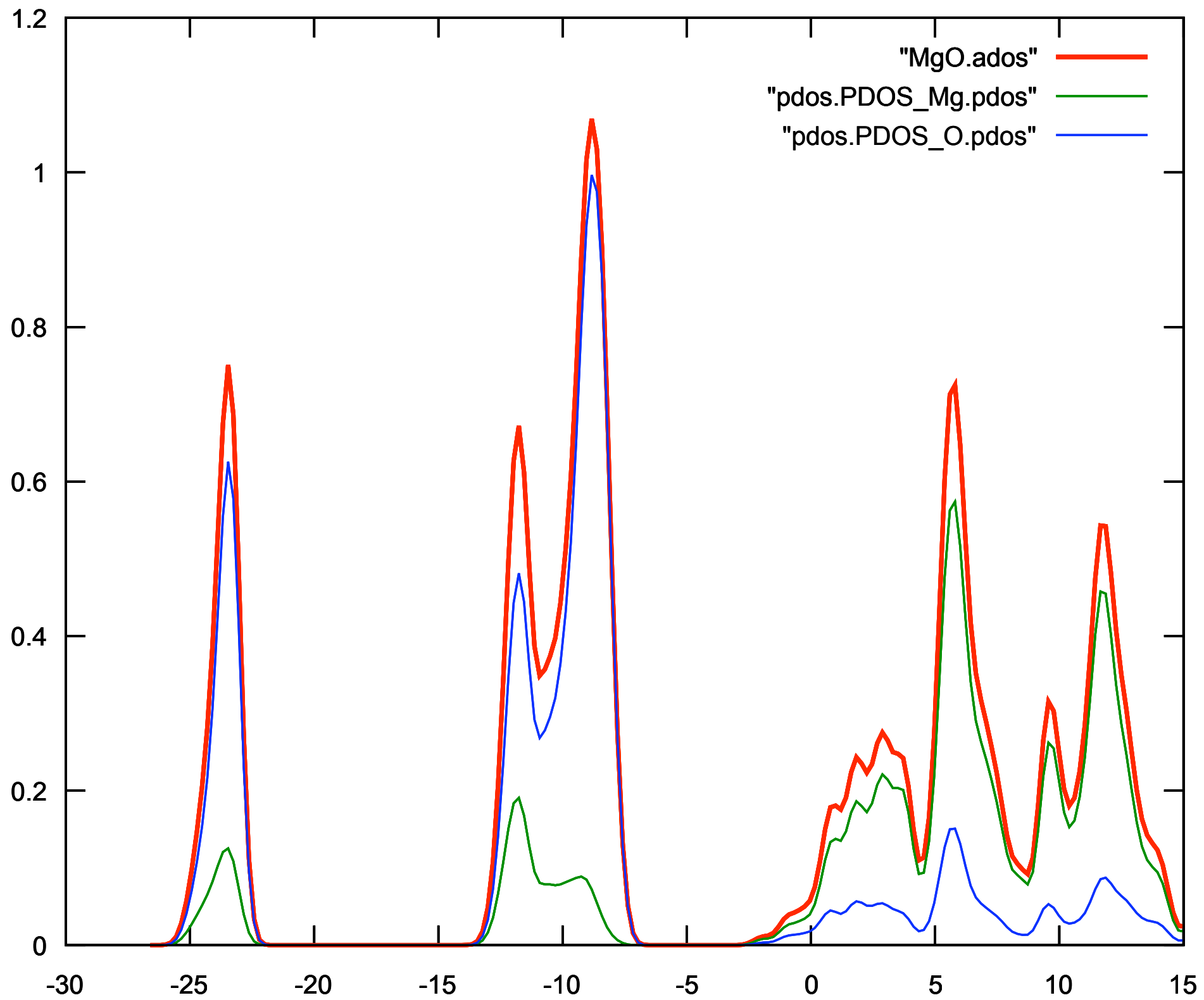
pdos.mpr

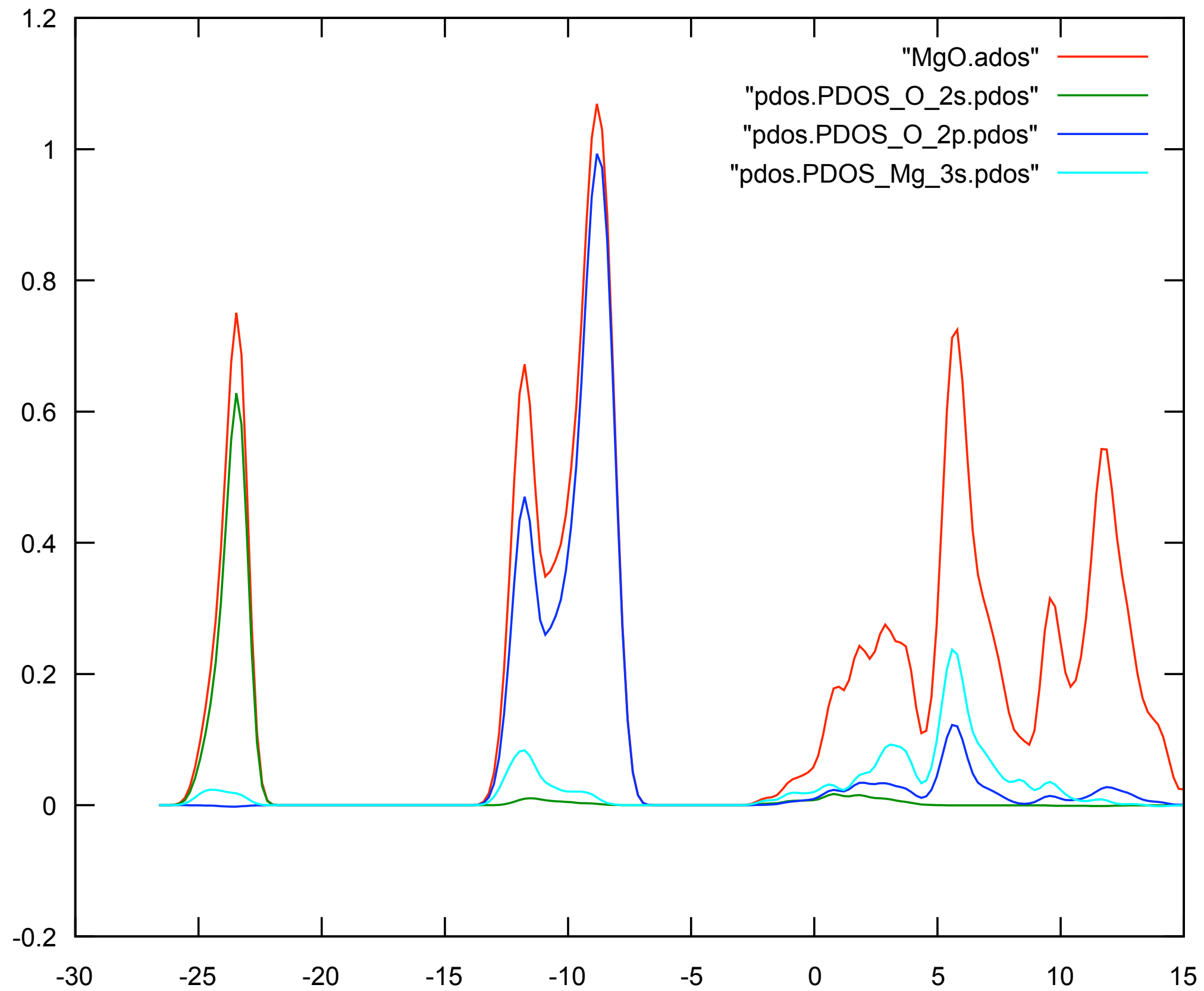
mprop -m Min\_Energy -M Max\_Energy input\_label

mprop -m -26.0 -M 15.0 pdos

-rw-r-----	1	ag	ag	10624	Jun	11	23:00	MgO.ados
-rw-r-----	1	ag	ag	6840	Jun	11	23:00	pdos.PDOS_Mg.pdos
-rw-r-----	1	ag	ag	6840	Jun	11	23:00	pdos.PDOS_Mg_3s.pdos
-rw-r-----	1	ag	ag	6840	Jun	11	23:00	pdos.PDOS_0.pdos
-rw-r-----	1	ag	ag	6840	Jun	11	23:00	pdos.PDOS_0_2p.pdos
-rw-r-----	1	ag	ag	6840	Jun	11	23:00	pdos.PDOS_0_2s.pdos

Total DOS (in specified range): SystemLabel.ados  
Projected DOS: InputLabel.CurveLabel.pdos





Density of states projected on orbital  $\mu$ :

$$g_{\mu}(\epsilon) = \sum_i \sum_{\nu} c_{\mu i} c_{i \nu} S_{\mu \nu} \delta(\epsilon - \epsilon_i)$$

Crystal Orbital Overlap Population (COOP)

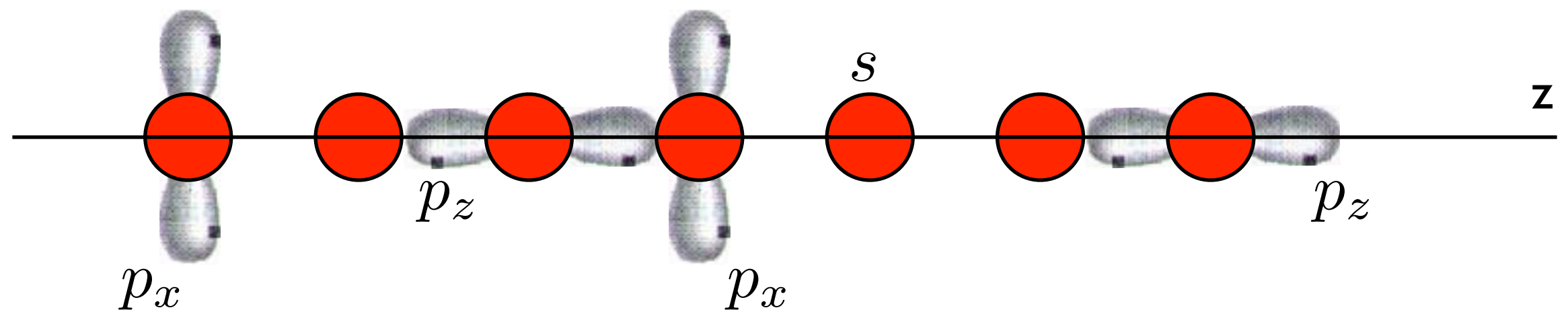
$$g_{\mu \nu}(\epsilon) = \sum_i c_{\mu i} c_{i \nu} S_{\mu \nu} \delta(\epsilon - \epsilon_i)$$

Crystal Orbital Hamilton Population (COHP)

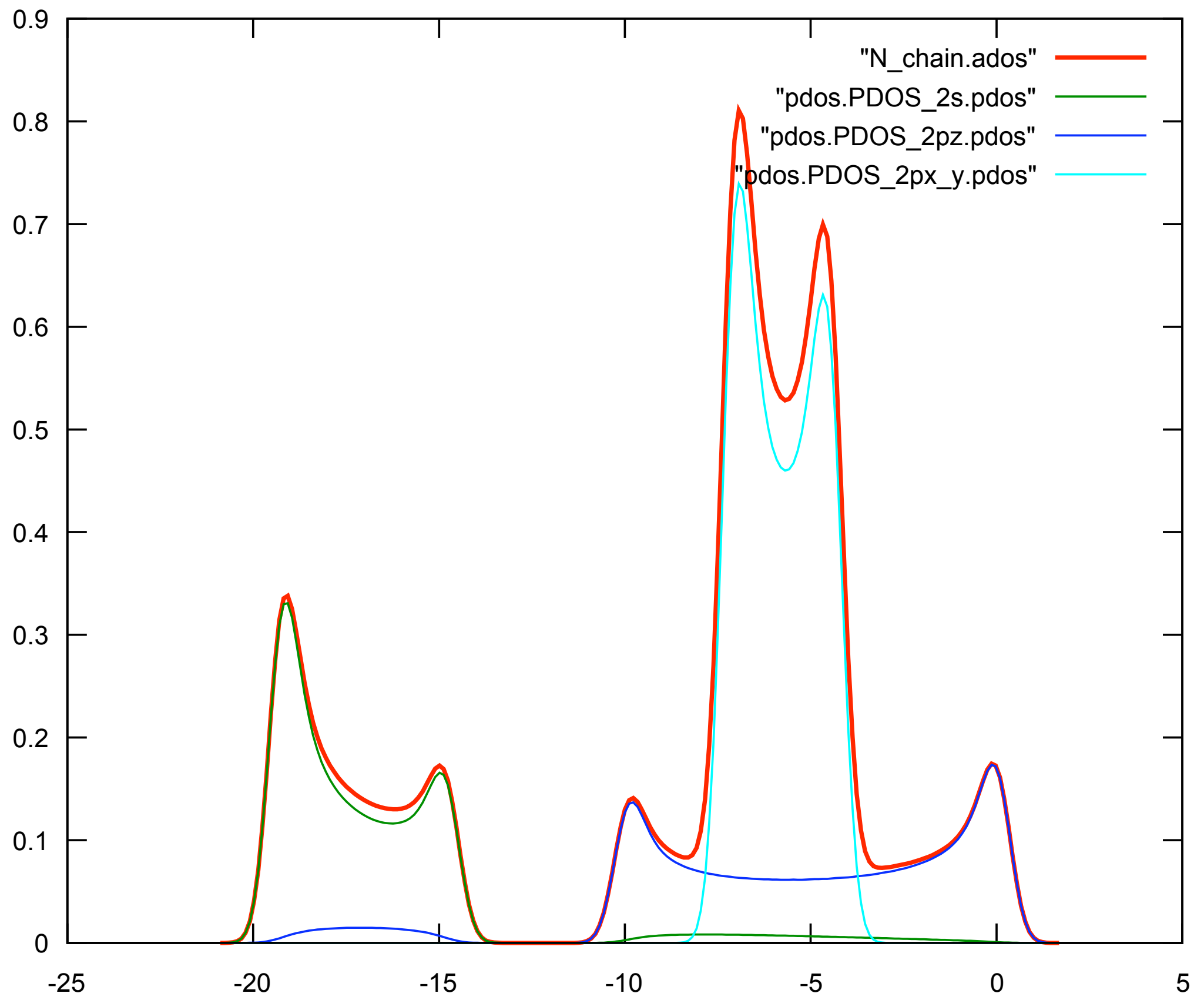
$$h_{\mu \nu}(\epsilon) = \sum_i c_{\mu i} c_{i \nu} H_{\mu \nu} \delta(\epsilon - \epsilon_i)$$

$$\psi_i(\mathbf{r}) = \sum_{\mu} \phi_{\mu}(\mathbf{r}) c_{\mu i},$$

# A chain of nitrogen atoms



(lattice constant: 2 Å)



**n\_chain**

**COOP**

N2s-N2px

N\_2s

1.95 2.05

N\_2px

N2px-N2px

N\_2px

1.95 2.05

N\_2px

N2s-N2pz

N\_2s

1.95 2.05

N\_2pz

N2s-N2s

N\_2s

1.95 2.05

N\_2s

N2pz-N2pz

N\_2pz

1.95 2.05

N\_2pz

Curve Label, orb1 spec,  
distance range, orb2 spec.

....

....

n\_coo.mpr

Distance ranges select nearest neighbors

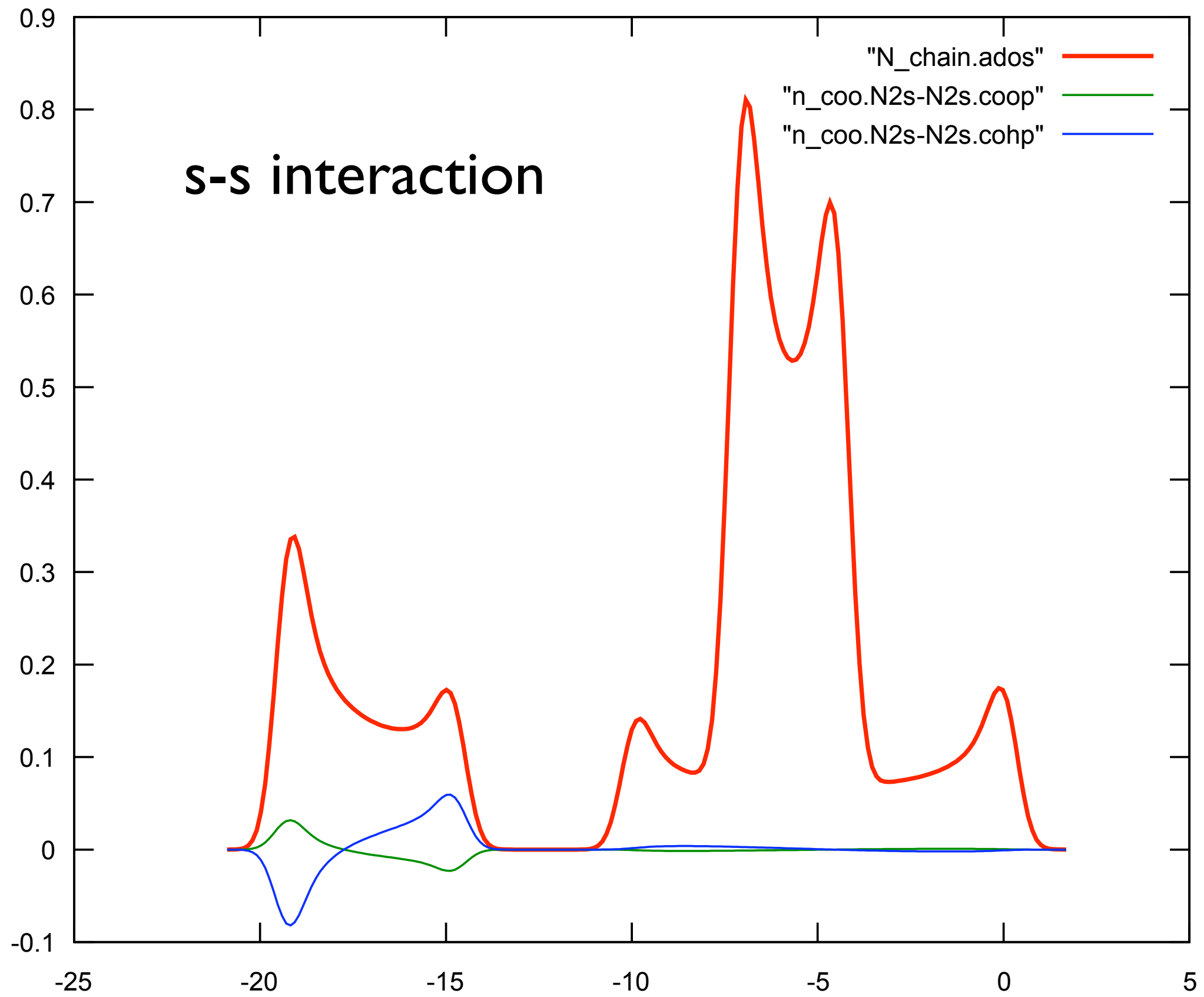


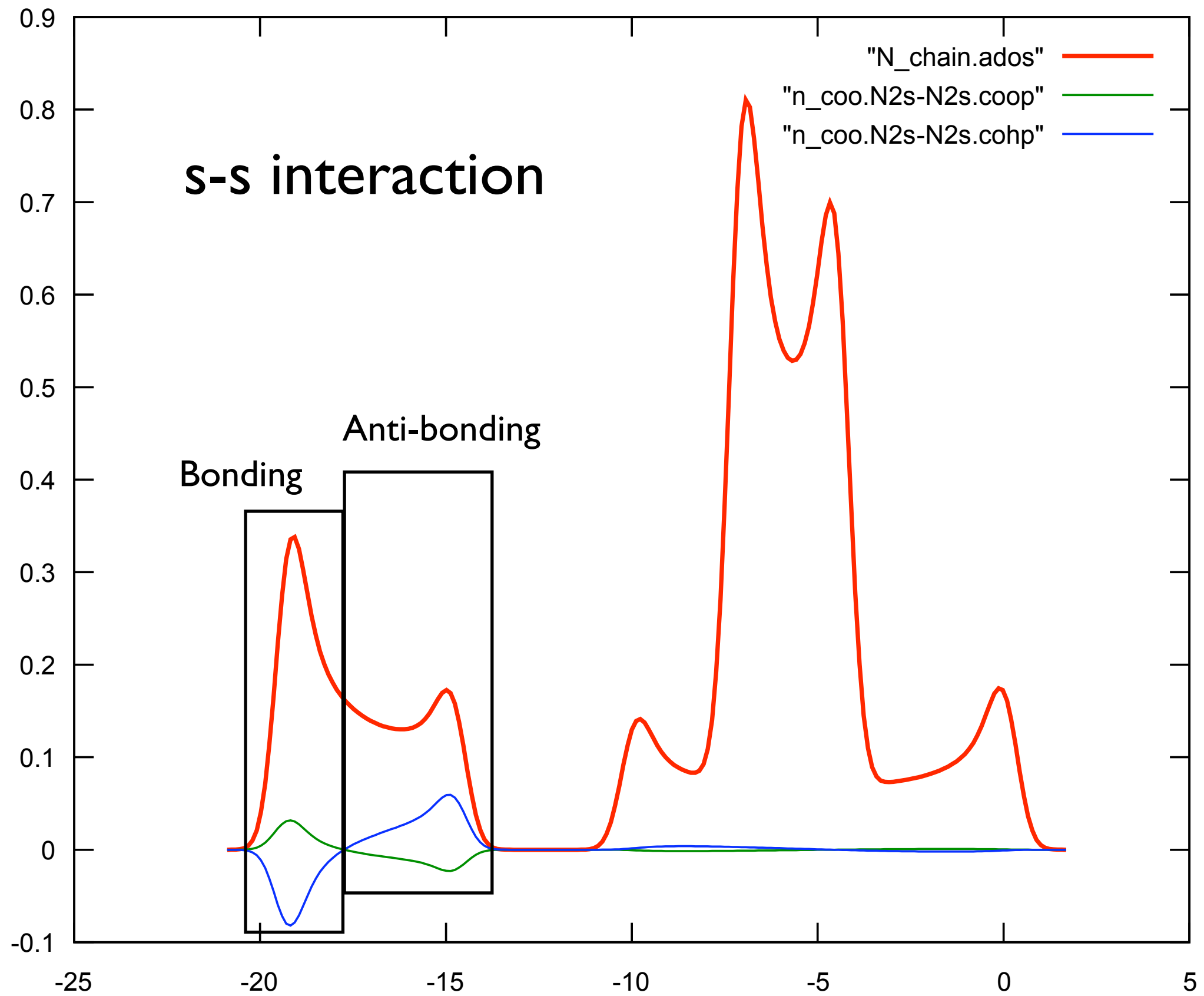
mprop n\_coo

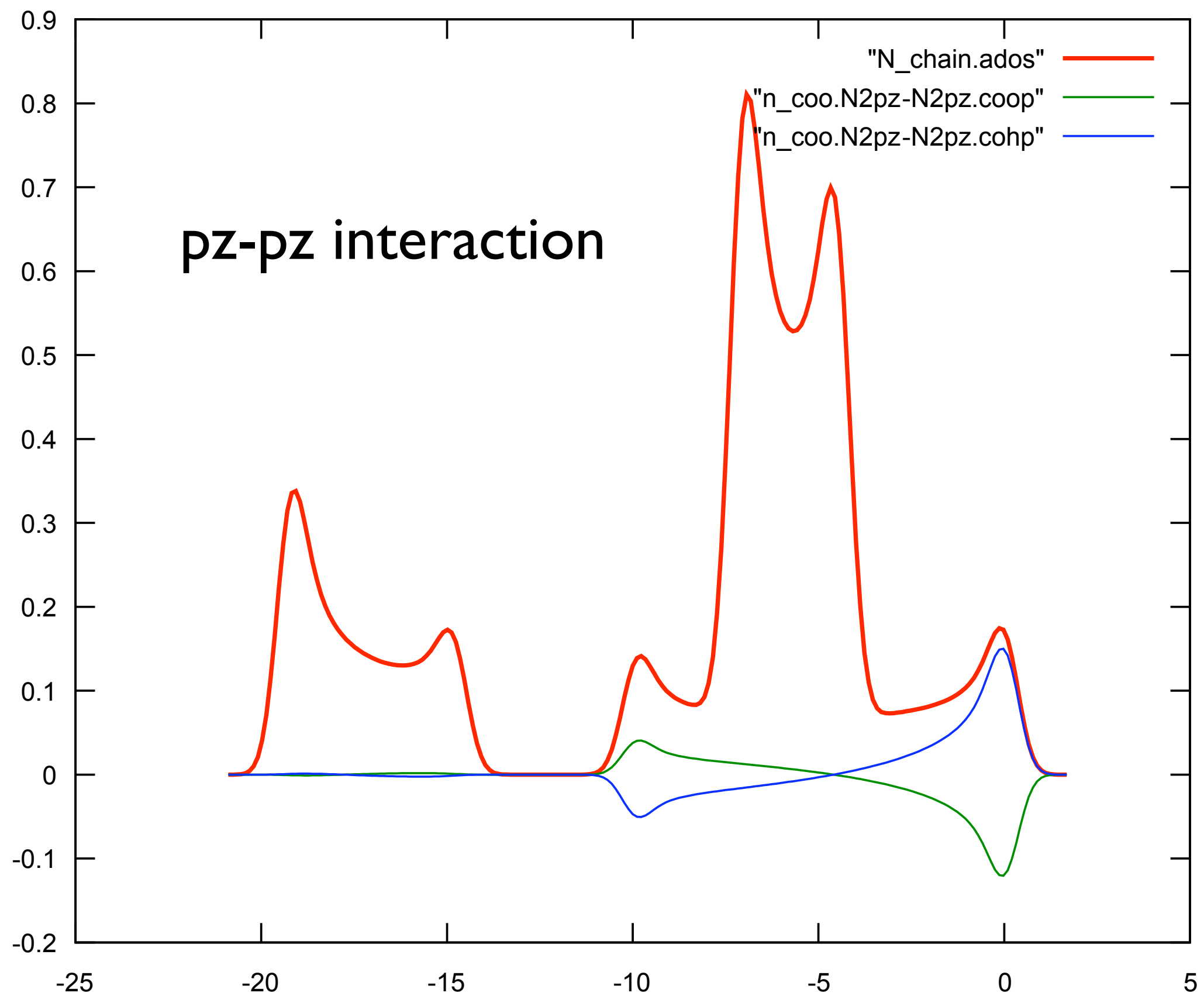
```
-rw-r----- 1 ag ag 6840 Jun 12 00:10 n_coo.N2px-N2px.coop
-rw-r----- 1 ag ag 6840 Jun 12 00:10 n_coo.N2pz-N2pz.coop
-rw-r----- 1 ag ag 6840 Jun 12 00:10 n_coo.N2s-N2px.coop
-rw-r----- 1 ag ag 6840 Jun 12 00:10 n_coo.N2s-N2pz.coop
-rw-r----- 1 ag ag 6840 Jun 12 00:10 n_coo.N2s-N2s.coop
```

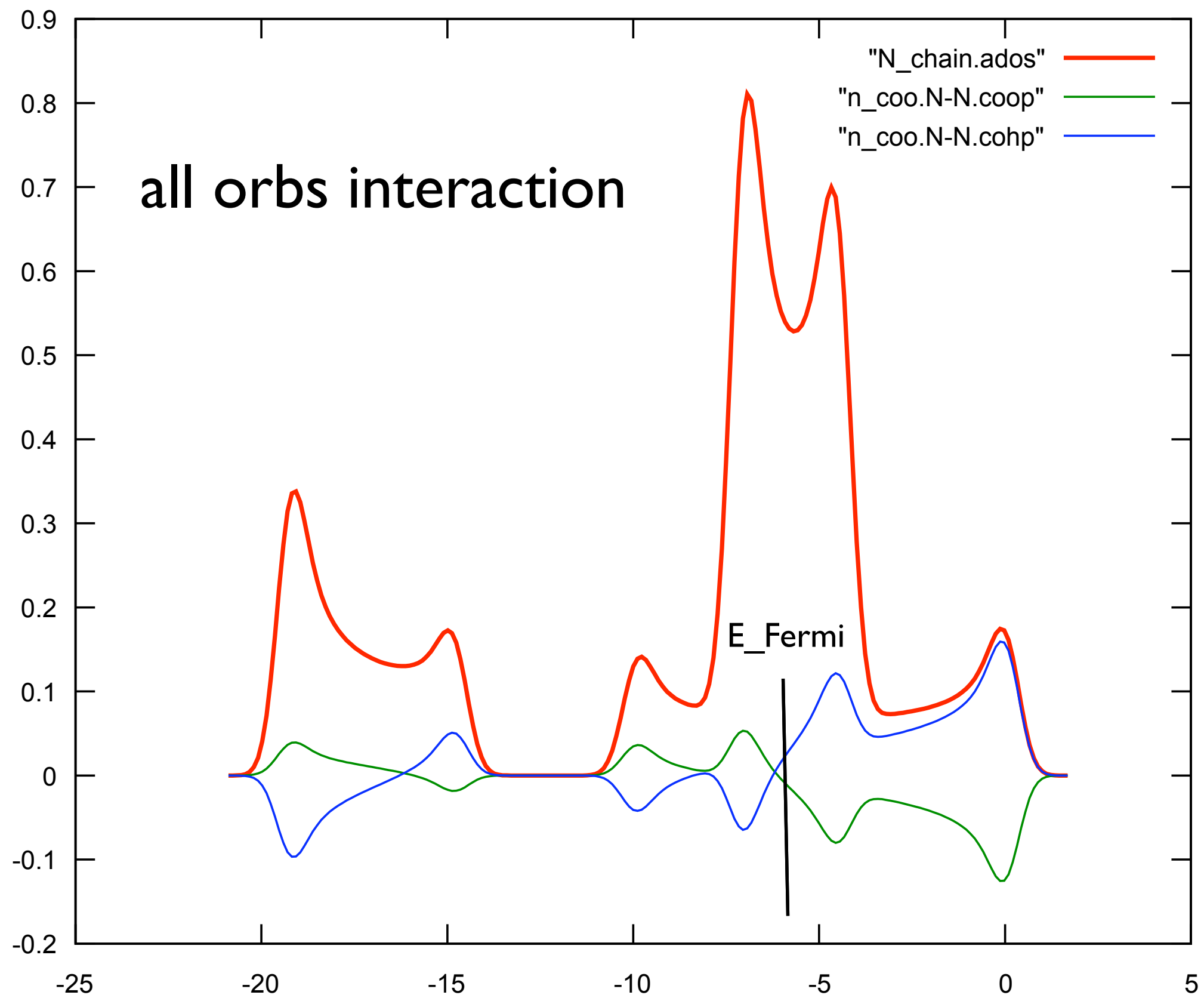
COOP file: InputLabel.CurveLabel.coop

COHP file: InputLabel.CurveLabel.cohp









# mprop lives in Util/COOP in the Siesta distribution

**mprop -h** Built-in help

\* MPROP PROGRAM

Miquel Llunell, Universitat de Barcelona, 2005

Alberto Garcia, ICMAB-CSIC, 2007

MPROP calculates both DOS projections and COOP curves using output files obtained with SIESTA. The atomic orbital (AO) sets are defined in an input file (MLabel.mpr).

Usage: mprop [ options ] MPROP\_FILE\_BASENAME

Options:

	-h:	print manual
	-d:	debug
	-l:	print summary of energy information
-s SMEAR	:	set value of smearing parameter (default 0.5 eV)
-m Min_e	:	set lower bound of energy range
-M Max_e	:	set upper bound of energy range

\* .mpr FILE STRUCTURE

```
    SLabel          # Name of the siesta output files
    DOS/COOP        # Define the curve type to be calculated
/-[ If DOS selected; as many blocks as projections wanted ]
|   projection_name # DOS projection name
\-   Subset of AO (*) # Subset of orbitals included
/-[ If COOP selected; as many blocks as projections wanted ]
|   curve_name      # COOP curve name
|   Subset I of AO (*) # Reference atoms or orbitals
|   d1 d2           # Distance range
\-   Subset II of AO (*) # Neighbour atoms or orbitals
(*) See below how to define subsets of AO
A final line with leading chars ---- can signal the end of the input
```

\* SUBSET OF AO USING ATOM\_SHELL NOTATION

List of atoms and shell groups of AO

General notation: ATOM\_SHELL

> ATOM: Atomic symbol refers to all the atoms of that type  
Integer number refers to the N-th atom in unit cell

> SHELL: Integer1+Letter+Integer2

> Integer1 refers to the n quantum number

> Letter refers to the l quantum number (s,p,d,f,g,h)

> Integer2 refers to a single AO into the n-1 shell

Alternatively, alphanumerical strings can be used

p-shells	1	y	d-shells	1	xy	4	xz
	2	z		2	yz	5	x <sup>2</sup> -y <sup>2</sup>
	3	x		3	z <sup>2</sup>		

Particular cases:

> Just ATOM is indicated: all the AO of the atom will be included

> No value for Integer2: all the AO of the shell will be included

Example: Ca\_3p Al 4\_4d3 5 O\_2py



- mprop can be used off-line to analyze as many curves and energy ranges as needed.
- It can replace the PDOS functionality within Siesta.
- It has a flexible orbital specification syntax.

# Useful reference with examples of the use of COOP and COHP curves

